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2017 CHEMISTRY EXAM TSSM 2017

Page 3 of 13 Question 4 Answer: C

Explanation: $3\text{KOH (aq)} + \text{AlCl}_3 \text{ (aq)}$

$3\text{KCl (aq)} + \text{Al(OH)}_3 \text{ (s)}$ $3\text{Zn(NO}_3)_2$

$\text{(aq)} + 2\text{K}_3\text{PO}_4 \text{ (aq)}$ $\text{Zn}_3 \text{ (PO}_4)_2 \text{ (s)} +$

$6\text{KNO}_3 \text{ (aq)}$ $\text{Sr(OH)}_2 + 2\text{NH}_4\text{CH}_3$

COO (aq) $2\text{NH}_4\text{OH} + \text{Sr(CH}_3\text{COO)}_2$

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(aq) Na_2SO_4 (aq) + $\text{Pb}(\text{NO}_3)_2$ (aq)
 PbSO_4 (s) + 2NaNO_3 (aq) Question 5

CHEMISTRY - St Leonard's College

Explanation: Use kinematics equations to find the initial vertical velocity, given $v_{\text{vert}} = 0$ at 1.8 sec after launch: $v = u + at$

$0 = u - 10 \times 1.8$ $u = 18 \text{ms}^{-1}$. Then, use

trigonometry to determine v at 35° : $v = 18$

$\cos(55) = 31.4 \text{ms}^{-1}$. Question 18. Answer: $R = 46 \text{ m}$ Explanation: $v_{\text{horiz}} = v \times \cos(35) =$

25.7ms^{-1} . $x = v_{\text{h}} \times t = 46.26 \text{ m}$.

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Units 3 & 4 – Written examination

(TSSM 's 2015 trial exam updated for the current study design) SOLUTIONS . PHYS EXAM TSSM 2017 Page 2 of 17 Section A - Multiple Choice (1 mark each) Question 1 Answer: C Explanation: The density of lines at a specific location in space reveals information about the strength of ...

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2011 Trial Examination SOLUTIONS

SECTION A – Core Area of Study 1 –

Motion ... 2011 PHYS EXAM 2 TSSM 2011

Page 5 of 21 Question 12 Answer: 6.7×10^4

N m⁻¹ Explanation: The gravitational

potential energy at 5 m ($mgh = 0.15 \times 10 \times 5$

$= 7.5$ J) is all converted to kinetic ...

Physics 2011 Exam 2 Unit 2 Solutions

2010 Trial Examination SOLUTIONS

SECTION A – Core Area of Study 1-

Electric Power Question 1 Answer: B ... 2010

PHYS EXAM 2 TSSM 2010 Page 3 of 16

Question 10 Answer: 300 turns Explanation:

Current at the generator is $10000/250 = 40$

A. Current in the lines is 10 A. Thus the

transformer is a step-up in the ratio of 1:4 ...

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TSSM 2018 Page 1 of 1 2018 Trial

Examination Letter STUDENT NUMBER

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Structure of book Section Number of questions Number of questions to be answered Number of marks A 20 20 20 B 6 6 66 Total 86 Students are ...

TSSM-Chemistry-2018-Unit-1-Trial-Exam.pdf - THIS BOX IS ...

2017 CHEMISTRY EXAM TSSM 2017
Page 2 of 13 SECTION A: Multiple-choice questions (1 mark each) Question 1 Answer: D Explanation: Ice is less dense than water as the regular array of water molecules formed as a result of hydrogen bonding causes gaps to form. Question 2 Answer: C Explanation: $2140 \text{ J} = 2.14 \text{ kJ}$ $n(\text{water}) = 2.14/6.0 = 0.357$
 mol $m(\text{water}) = 0.357 \times 18.0 = 6.42 \text{ g}$
Question 3 Answer: D ...

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TSSM 2011 Page 1 of 19 PHYSICS Unit 1

– Written examination 1 2011 Trial

Examination SOLUTIONS SECTION A

Area of Study 1 – Nuclear Physics and

Radioactivity Question 1 Answer:A

Explanation: As per definition of each type of radioactive decay. Alpha is the heaviest and doubly

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