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?  $4\text{CO}_2 + 5\text{H}_2\text{O}$ . In products we have; 13 O atoms, there must be 13 O atoms in reactants. But we have 10, so "z" must be 3. In products we have 4 C atoms, to balance it there must be 4 in reactants, so "x" is 4.

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fundamental particles protons (+1 charge), neutrons (zero charge), and electrons (-1 charge).

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## **Equal amounts of two acids—HCl and HCO<sub>2</sub>H (formic acid ...**

In addition, a recent tenure as a mentor of the U.S. Chemistry Olympiad Team, Professor Kotz has received numerous honors, including a State University of New York Chancellor's Award (1979), a National Catalyst Award for Excellence in Teaching (1992), the Estee Lectureship in Chemical Education at the University of South Dakota (1998), the Visiting Scientist Award from the Western Connecticut Section of the American Chemical Society (1999), and the first annual Distinguished Education Award ...

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